

A Novel Periselective Cycloaddition of Cycloheptatriene with Cyclohexa-2,4-Dienones

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A highly periselective cycloaddition of cycloheptatriene and cyclohexa-2,4-dienones (2a-c) to *endo* tricyclic systems (6a-c) and their photo conversion to cage compounds (7a-c) has been reported. It has been shown that the adduct 6c arise *via* Cope rearrangement of the initially formed $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ adduct 5c.

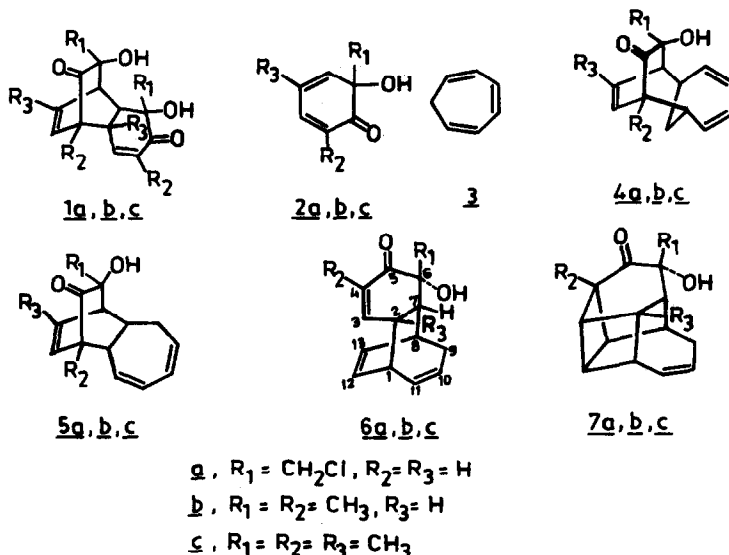
There has been a resurgence of interest in the pericyclic reactions between complex conjugated polyenes such as tropone, fulvenes, cycloheptatriene, oxapines, and azapines, in order to explore higher order cycloadditions and their potential for rapid creation of complex carbocyclic systems.^{1,2} Most of the studies in this area, however, have been restricted to molecular systems selected for examining mechanistic and theoretical aspects. Synthetic applications of pericyclic reactions between polyene addends have been limited due to multiple modes of cycloadditions and lack of periselectivity.³

In context with a synthetic endeavour, we required the tricyclic system 5 which we thought to assemble *via* cycloaddition between cycloheptatriene (CHT) and cyclohexa-2,4-dienone 2 (dienone). This thought was based on the observation that cyclohexa-2,4-dienones of type 2 react as 4π addend during their cycloaddition with dienes and as well as trienes.⁴ At the outset, however, it was clear that cycloheptatriene (CHT) may undergo $\pi^{6s}(\text{CHT}) + \pi^{4s}(\text{dienone})$, $\pi^{4s}(\text{CHT}) + \pi^{2s}(\text{dienone})$ and $\pi^{2s}(\text{CHT}) + \pi^{4s}(\text{dienone})$ type of cycloadditions, leading to a variety of products (4-6) (Scheme-1).

We wish to report herein a novel, highly periselective and synthetically useful cycloaddition of cycloheptatriene 3 with cyclohexa-2,4-dienones 2a,b and c leading to adducts 6a,b and 5c, 6c respectively. We also report that the adduct 6c is formed *via* Cope rearrangement of the initially formed $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ adduct 5c (Scheme 1). A facile photochemical $\pi^{2s} + \pi^{2s}$ cycloaddition of 6a-c to the corresponding cage compounds 7a-c is also reported. To the best of our knowledge, this is the first report on the cycloaddition of cycloheptatriene and cyclohexadienones.

Generation of cyclohexadienone 2a by pyrolysis⁴ (160°C) of the diol-dimer 1a and subsequent trapping with cycloheptatriene gave an adduct 6a as exclusive product in excellent yield (82.5%). The structure of the adduct was revealed from the following data, m.p. 111°C, IR ν_{max} : 1690 cm^{-1} , ¹H NMR (300MHz, CDCl₃) δ : 6.53 (dd, $J_1=10\text{Hz}$, $J_2=3\text{Hz}$, 1H, H₃, proton at β -carbon of α,β -enone), 6.15 (dd with long range

couplings, $J_1=10\text{Hz}$, $J_2=3\text{Hz}$, 1H,H4, proton at α -carbon of α,β -enone), 6.02 (dd, $J_1=J_2=8\text{Hz}$, 1H, H12/H13), 5.92(dd, $J_1=J_2=8\text{Hz}$, 1H, H13/H12), 5.85(m of dd, $J_1=12\text{Hz}$, $J_2=9\text{Hz}$, $J_3=3\text{Hz}$, 1H, H11) and 5.48(t of d, $J_1=12\text{Hz}$, $J_2=4\text{Hz}$, 1H, H10), 4.2(s, 1H, OH), 3.7(ABq, $J=12\text{Hz}$, 2H, CH₂Cl), 3.44(d of dd, $J_1=8\text{Hz}$, $J_2=3\text{Hz}$, 1H, H2), 2.92 (d, $J=8\text{Hz}$, 1H, H7), 2.78 (m, 2H) and 2.2 (m, 2H). These assignments were made with the help of



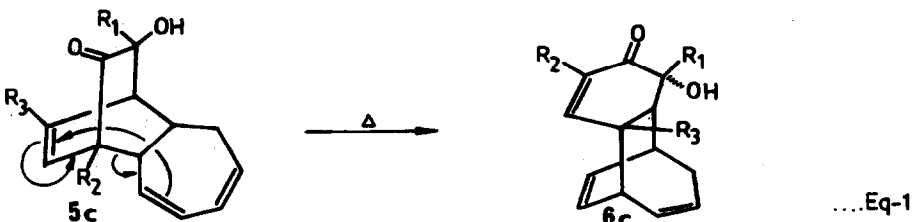
Scheme-1

cosy experiment. The ¹³C NMR of the adduct showed signals at δ 197.38(CO), 152.5, 134.1, 129.5, 127.2, 126.4 (olefinic carbons), 77.1(CH₂Cl), 52.8, 46.4, 45.3, 38.3, 35.5 and 28.6 for methine and methylene carbons. The *endo* stereochemistry of **6a** was confirmed through its facile photoconversion to the cage compound **7a** upon irradiation (200W, Hanovia, mercury vapour lamp) in acetone. Similarly the reaction of the diol-dimer **1b** with cycloheptatriene gave a single *endo* adduct **6b**⁵ which also underwent $\pi^{2s} + \pi^{2s}$ photoreaction to the corresponding cage molecule **7b**⁵ in excellent yield (84.5%).

It was surprising that the alternate modes of symmetry allowed^{6,7} cycloadditions were not observed, especially since cyclohexadienones are well-known⁴ to react exclusively as 4π addend during their cycloaddition with dienes/ trienes. In order to understand the observed periselectivity the coefficients of frontier molecular orbitals of **2** and **3** were examined, however it did not lead to any fruitful correlation.

Therefore it was considered⁸ that the adducts **6a,b** may arise *via* Cope rearrangement of the initially formed $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ adducts **5a,b** whose formation is in agreement with FMO consideration. However, the attempts to isolate **5a,b** were

unsuccessful. In this context, therefore we attempted the cycloaddition of 2c with 3. Indeed, the pyrolysis (150°C) of 1c with 3 gave the $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ adduct 5c (m.p. 123°C, IR ν_{max} : 1720 cm^{-1}) in addition to the enone 6c,⁹ the latter arising due to tandem $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ cycloaddition and a 3,3-shift, since the adduct 5c was found to undergo a 3,3-shift to give 6c (IR, NMR) upon further heating (150-160°C) (eq. 1). In view of this, it appears that the compounds 6a,b might also be formed following a similar pathway.¹⁰



In summary, we have explored the cycloaddition of cyclohexa-2,4-dienones with cycloheptatriene and demonstrated that the $\pi^{4s}(\text{CHT}) + \pi^{2s}(\text{dienone})$ adducts arise *via* an tandem $\pi^{4s}(\text{dienone}) + \pi^{2s}(\text{CHT})$ addition and Cope rearrangement.

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- Data for 7a, m.p. 141°C. IR ν_{max} : 1720 cm^{-1} , ^1H NMR (300MHz, CDCl_3) δ : 5.8 (m of d $J=12\text{Hz}$, 1H), 5.5 (m of dd, $J_1=12\text{Hz}$, $J_2=18\text{Hz}$, 1H), 3.6 (part of AB q, $J=12\text{Hz}$, 1H, CH_2Cl), 3.42 (part of ABq, $J=12\text{Hz}$, 1H, CH_2Cl), 3.40 (m, 1H, CH), 3.13 (s, 1H, OH), 3.05 (m, 2H, CH), 2.85 (m, 3H, CH), 2.65 (dt of part of ABq, $J_1=18\text{Hz}$, $J_2=$

6Hz, $J_3=2\text{Hz}$, 1H,CH₂), 2.5(q of part of ABq, $J_1=18\text{Hz}$, $J_2=3\text{Hz}$,1H,CH₂) and 2.30 (m,1H,CH); ¹³C NMR(75MHz): δ 214.3, 132.4, 127.0, 78.85, 47.47, 46.7, 46.2, 44.6, 42.8, 42.37, 38.65, 37.8, 35.4 and 32.2. Analysis: Found, C, 67.3; H, 6.4%. Calcd. for C₁₄H₁₅O₂Cl, C, 67.06; H, 5.98%.

Data for compound 6b, liquid, IR ν_{max} : 1690 cm⁻¹, ¹H NMR(300MHz,CDCl₃): δ 6.15(m,1H), 5.95(d of dd, $J_1=J_2=8\text{Hz}$, $J_3=1.5\text{Hz}$,1H), 5.88(dd, $J_1=J_2=8\text{Hz}$, 1H), 5.82 (m,1H), 5.45(t of d, $J_1=10\text{Hz}$, $J_2=3.5\text{Hz}$,1H), 4.0(br s,1H),3.38(complex m,1H),2.81 (d, $J=8\text{Hz}$,1H), 2.7(m,2H), 2.17(m,2H), 1.8(dd, $J_1=2.5\text{Hz}$, $J_2=1.25\text{Hz}$,3H,CH₃), 1.2(s, 3H,CH₃),. ¹³C NMR(75MHz): δ 202.5, 147.3, 134.2, 132.9, 130.7, 129.4, 127.5, 74.5, 49.6, 44.6, 38.9, 35.8, 29.7, 28.49, 15.45. Analysis: Found, C, 78.4; H,8.0% Calcd. for C₁₅H₁₈O₂, C, 78.26; H, 7.82%.

Data for compound 7b, m.p. 110°C,IR ν_{max} : 1720 cm⁻¹, ¹H NMR(90MHz,CDCl₃) δ 5.5(m,2H,olefinic H), 3.45(m,1H,CH), 3.2-2.7(br 4H,CH), 2.6-2.3(m,5H,CH and OH), 1.22(s,3H,CH₃),and 1.20(s,3H,CH₃). ¹³C NMR(75MHz): δ 218.8,132.3,127.2, 76.5,51.98,47.27,46.1,45.6,43.3,37.8,36.2,35.6,22.2, and 15.1. Analysis: Found, C, 78.4; H, 8.0%. Calcd. for C₁₅H₁₈O₂, C, 78.26; H, 7.82%.

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8. We are highly grateful to the referee for this suggestion.

9. Data for compound 5c, m.p. 123°C. IR ν_{max} : 3450, 1720 cm⁻¹, ¹H NMR(500MHz, CDCl₃): δ 6.0(m,1H), 5.9-5.8(m,3H), 3.45(d, $J=10\text{Hz}$,1H), 2.68(t, $J=2\text{Hz}$,1H), 2.6(d of t, $J_1=10\text{Hz}$, $J_2=3\text{Hz}$,1H), 2.18-2.05(complex m,3H), 1.92(d, $J=1.5\text{Hz}$,3H,CH₃), 1.25(s, 3H,CH₃),and 1.15(s,3H,CH₃). Analysis: Found, C, 78.43; H, 8.22%. Calcd. for C₁₆H₂₀O₂, C, 78.68; H, 8.19%.

Data for compound 6c, liquid, IR ν_{max} :3480,1690 cm⁻¹, ¹H NMR(500MHz,CDCl₃) δ 6.05(dd, $J_1=J_2=8\text{Hz}$, 1H), 5.9(br s, 1H), 5.8(dd, $J_1=J_2=8\text{Hz}$, 1H), 5.75(t of dd, $J_1=J_2=8.5\text{Hz}$, $J_3=3\text{Hz}$, 1H), 5.46(t of d, $J_1=11\text{Hz}$, $J_2=3.5\text{Hz}$, 1H), 2.92(m,1H), 2.33 (s,1H), 2.31(d, $J=8\text{Hz}$, 1H), 2.18(m,2H),1.75(s,3H,CH₃), 1.45(s,6H, 2 x CH₃). This compound was irradiated to give 7c, m.p. 120°C, IR ν_{max} : 3480, 1710 cm⁻¹ ¹H NMR(300MHz, CDCl₃): δ 5.6(complex m, 2H), 3.5(dd, $J_1=15\text{Hz}$, $J_2=10\text{Hz}$, 1H) 2.82(s,1H,OH), 2.6(m,1H), 2.45(complex, m,2H), 2.3(complex, m, 4H), 1.2(s,3H, CH₃), and 1.0(s, 6H, 2x CH₃). ¹³C NMR(75MHz): δ 219.22(CO),132.21,125.97, 50.9, 48.7, 47.41, 44.81, 44.0, 36.23, 35.71, 27.25, 25.43, and 15.29.

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